Course outcomes of B.Sc. 1st(chemistry) Semester 1st Course physical chemistry 1.1 Gaseous state CO1: To develop the concept of gases CO2: To understand the behaviour of gases CO3: To develop the concept of vanderwal l equation Course physical chemistry 1.2 Critical phenomena CO1: To learn about liquification of gases CO2: To develop the knowledge of gases CO3: To learn about real gases Course physical chemistry 1.3Liquid state CO1: To understand the power of liquid motion CO2: To understand the basics of liquid CO3: To increase the knowledge of viscosity Course physical chemistry 1.4 Solid state CO1: To understand the structures of solids CO2: To increase the knowledge about construction of buildings CO3: To find the concept of glasses mobility Semester 2<sup>nd</sup> Course physical chemistry 2.1(chemical kinetics) CO1: To learn how to calculate time consumed in reactions CO2: To understand the effects of temperature on reactions CO3: To understand the speed of reactant Course physical chemistry 2.2 Electrochemistry CO1: To understand the production of current from chemicals CO2: To learn about conductance of ions

CO3: To enhance the knowledge about batteries Courseinorganic chemistry 1.1 Atomic structure Sem. 1st CO1: To understand the importance of atom CO2:To learn about structure of universe CO3: To understand the 3D model of atom Course inorganic chemistry 1.2 Covalent bonding CO1: To understand the nature of bonds CO2: To learn about structure of compounds CO3:To understand the resonating structure Semester 2<sup>nd</sup> co Course inorganic chemistry 2.1Hydrogen bonding and vander wall force CO1: To understand the bonding of electro negative atoms CO2: To learn about the concept of low density of ice CO3: To develop the concept of high boiling of water Course inorganic chemistry 2.2 Metallic bond and semiconductor CO1: To understand the value of metals CO2: To learn about the structures of semiconductor CO3: To enhance the effectiveness of our apparatus Course inorganic chemistry 2.3 S Block elements CO1: To know about reactive metals CO2: To learn about behaviour of sodium like metals CO3: To understand the difference between soft and hard metals

Course inorganic chemistry 2.4Chemistry of Nobel gases

CO1: To increase the knowledge of noble gases

CO2: To learn about the reactivity of xenon

CO3: To find medicinal effect of radon in cancer treatment

Course inorganic chemistry 2.5 P block elements

CO1: To understand the nature of non metals

CO2: To learn about the conducting behaviour of carbon

CO3: To enhance the overall study of electro negative atoms

## Semester 1st

Course organic chemistry 1.1Structure and bonding CO1: To understand the resonance in compound s CO2: To learn about the stability of benzene CO3: To understand the structures compounds Course organic chemistry 1.2 steriochemistry CO1: To understand the effects of light on compounds CO2: To learn about the 3D structure CO3: To find difference between simple and Optical Course organic chemistry 1.3 Mechanism of organic reaction CO1: To understand the mechanism of reactions CO2: to learn about formation of new compounds CO3: To develop the concept of bond formation Course organic chemistry chemistry 1.4 Alkane and cycloalkane CO1: To learn about alkane CO2: To understand the nature of gases CO3: To develop the concept of polarity Semester 2<sup>nd</sup> Course organic chemistry 2.1Alkane CO1: To understand the nature of alkane CO2: To learn about markonikoves CO3: To understand about sytzeff rule Course organic chemistry 2.2 Arene and aromaticity CO1: To understand Huckel's rule CO2: To learn about benzene and its derivatives CO3: To learn about aromaticity

Course organic chemistry 2.3 Diene and alkyne

- CO1: To understand about diane
- CO2: To learn about alkyne
- CO3: To understand the nature of diane
- Course organic chemistry 2.4 Alkyl and Aryl halide
- CO1: To understand the effects of alkyl group on alkane
- CO2: To understand the nature of aryl halide
- CO3: To learn about these